## SODIUM HYDROXIDE KIT





QUANTITY	CONTENTS	CODE	
30 g	*Barium Chloride Powder	*6073-G	*Reagent is a potential health hazard. <b>READ SDS</b> : lamotte.com <b>Emergency information</b> : Chem-Tel USA 1-800-255-3924 Int'l, call collect, 813-248-0585
2 x 30 mL	*Hydrochloric Acid, 2.5N	*6251DR-G	
60 mL	Deionized Water	5115PT-H	
15 mL	*Phenolphthalein Indicator, 1%	*2246-E	
1	Pipet, 1.0 mL, plastic	0354	
1	Test Tube, 5-10-15-20-25-30 mL, plastic, w/cap	0715-DRT	
1	Direct Reading Titrator, 0 - 10 Range	0377	
1	Spoon, 0.5 g, plastic	0698	
To order inc	lividual reagents or test kit components,	-	

Read the LaMotte Direct Reading Titrator Manual before proceeding. The Titrator is calibrated in % Sodium Hydroxide. Each minor division equals 0.2%.

## **PROCEDURE**

specified code number.

- 1. Use the 1.0 mL pipet (0354) to add 1.0 mL of sample to test tube (0715-DRT).
- 2. Fill the test tube to 10 mL line with Deionized Water (5115).
- 3. Use the 0.5 g spoon (0698) to add one level measure of \*Barium Chloride Powder (6073). A white precipitate will form if carbonates are present.
- **4.** Add two drops of \*Phenolphthalein Indicator, 1% (2246). Cap and mix. Solution will turn pink.
- 5. Fill the Direct Reading Titrator (0377) with the \*Hydrochloric Acid, 2.5N (6251). Insert the tip of the Titrator into the center hole of the test tube cap.
- **6.** While gently swirling the tube, slowly press the plunger to titrate until the pink color disappears.
- 7. Read the concentration of the test sample where the plunger meets Titrator scale. Record as % Sodium Hydroxide.

**EXAMPLE:** Titrator reading is 3 minor divisions below line 7.

 $7 + (3 \text{ divisions } \times 0.2) = 7.6\% \text{ NaOH}$